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a chemical compound. Reveals the number of atoms

of each element contained in a single molecule ...

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Compounds SECTION 1 SHORT ANSWER Answer the
following questions in the space provided. 1. c In a
Stock system name such as iron(III) sulfate, the
Roman numeral tells us (a) how many atoms of Fe are
in one formula unit. (b) how many sulfate ions can be

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attached to the iron atom. (c) the charge on each Fe ion.

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Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO₃ a. copper(II) carbonate Na₂SO₃ b. sodium sulfite (NH₄)₃PO₄ c. ammonium phosphate SnS₂ d. tin(IV) sulfide HNO₂ e. nitrous acid 2.

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CHAPTER 5 REVIEW The Periodic Law SECTION 1
SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

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5 The Periodic Law

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